

Efficiency of applying Fe and Zn singly or in various combinations as nano and non-nano forms of foliar spray on snap bean (*Phaseolus Vulgaris*).

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Abstract

16 combinations of 4 Fe and 4 Zn foliar spray solutions of Fe₀, Fe₁, Fe₂ and Fe₃ and similars of Zn₀, Zn₁, Zn₂ and Zn₃ (mgL⁻¹) i.e. 0, 250-nano, 500-nano and 1000-micro (ordinary) FeL⁻¹ were sprayed on snap bean (*Phaseolus Vulgaris*). Sources were FeSO₄.7H₂O and ZnSO₄.7H₂O. Spraying was done on days 37, 44, 51 and 58 after seeding. Plants of Zn₀Fe₁ gave the lowest fresh yield of 652 kg ha⁻¹, while plants of Zn₂Fe₃ gave the highest yield of 1156 kg ha⁻¹ with an increase of 77.3% over the lowest. Applying Zn or Fe singly or various combinations proved of positive effect. A decrease in yields occurred by application of Zn and Fe at the high dose of 1000 mgL⁻¹ indicating a retarding effect at such high concentrations.

Keywords: nano fertilizer, snap bean, foliar spray.

Introduction

Application of nano fertilizers is a new way to increase nutrient use efficiency in crop production. Nano particles have extremely small size of few nanometers' diameter which have properties of high specific area and a high surface energy which render them highly efficient as fertilizers ((Lal, 2008 and Shebl et al., 2019).

Zinc (Zn) is an essential micronutrient for plant and iron (Fe) is another important micronutrient ; both are needed for proper healthy plant growth (Tilman et al., 2002; Mueller et al., 2012).

Within the framework of sustainable agriculture, applying nano fertilizers is one of the most promising approaches to insure increased crop production. DeRosa et al. (2010) considered fertilizer nanotechnology a potential solution for the low efficiency of fertilizer application.

Kalra et al. (2020) stressed the importance of micronutrients to plant and human health. Bioavailability of micronutrients occurs in about fifty percent of the world arable lands (Marschner 1986). Shang et al. (2019) stated that nano technology can enhance bioavailability of nutrients. In order to ensure agricultural practices which are safe in environment, nanotechnology can render slow or controlled release of fertilizer nutrients. Nanotechnology has gradually moved from the lab-based experimental trials to practical applications.

The current study aims at assessing foliar spray of nano Fe and Zn in comparison with non-nano (micro-) ordinary Fe and Zn on snap bean "Pole Bean" (*Phaseolus Vulgaris*) which one of the most important economic members of the *Fabaceae* crops in Egypt.

Materials and methods.

A field was carried out during the winter season of 2017 to assess foliar spray of nano Fe and Zn in comparison with non-nano (micro-) Fe and Zn on snap bean (*Phaseolus Vulgaris*). The experimental design was a randomized complete block factorial. **First Factor** Fe solutions of Fe₀, Fe₁, Fe₂ and Fe₃ of (mgL⁻¹) 0, 250-nano, 500-nano and 1000-micro (ordinary) FeL⁻¹ respectively.

Second factor: Zn solutions of Zn₀, Zn₁, Zn₂ and Zn₃ concentrations exactly as Fe.

Sources were FeSO₄.7H₂O and ZnSO₄.7H₂O. Spraying was done on 4 occasions: days 37, 44, 51 and 58 after seeding. Treatment combinations are 16 (4 Fe X 4 Zn) in triplicates with 10.5 m² plots. Seeds were sown (2 seeds per hill) on the 15th of February 2017 on one side of the ridge in hills 15-cm apart. Bean pods were harvested after 80 days of growth. The soil was a slightly alkaline sandy loam (Table 1)

Table 1. Properties of the 0-30-cm soil of the experimental field

Property	Value	Property	Value
Particle size distribution		Soluble ions (mmolL ⁻¹)	
Sand %	77.9	Ca ²⁺	6.6
Silt %	12.0	Mg ²⁺	3.5
Clay %	10.1	Na ⁺	10.8
Texture*	Sandy Loam	K ⁺	0.5
Saturation percent	35	Cl ⁻	10.4
Water holding capacity	25	SO ₄ ²⁻	5.9
		HCO ₃ ⁻	4.5
		CO ₃ ²⁻	0.0
Chemical properties		Available N	9

	 P	9
	 K	42
Organic matter (g kg ⁻¹)	5 Fe	9
	 Zn	5
CaCO ₃ (g kg ⁻¹)	51	* According to the International Soil Texture Triangle (Moeys, 2016).	
pH(1:25 w:v soil/ water)	7.5	** Extracts are: KCl (for N); DTPA and NH ₄ HCO ₃ . (For P, K, Fe and Zn).	
EC "paste extract" (dSm ⁻¹)	2.18		

Results and Discussion

Fresh yield of pods (Table 2)

Treatments receiving no Zn but low nano Fe (Zn₀Fe₁) gave the lowest yield of 652 kg ha⁻¹, while plants receiving high nano-Zn with high ordinary Fe (Zn₂Fe₃) gave the highest yield of 1156 kg ha⁻¹ with an increase of 77.3% over the lowest. This shows the importance of applying nano Zn along with iron in enough doses to obtain increased plant growth and yield. **Bastani et al (2018)** found that application of nano-Fe increased tobacco (*Nicotiana tabacum*) growth. **Karimi, et al (2014)** emphasized the need for applying Fe and Zn to plants. **Karimi et al. (2014)** obtained increased yields of mung beans (*Vigna Radiate* L.) upon applying nano-iron chelate as foliar application.

The main effect of Fe shows lowest given by Fe₀ and highest given by Fe₂ with a pattern of Fe₂ > Fe₁ ≥ Fe₃ ≥ Fe₀ and increases of 19.1, 3.5 and 1.7% given by Fe₂, Fe₁ and Fe₃ respectively over Fe₀.

The main effect of Zn fertilization shows lowest given by Zn₀ and highest given by Zn₁ with a pattern of Zn₁ ≥ Zn₃ ≥ Zn₂ > Zn₀; and increases of 19.2, 14.9 and 11.7% given by Zn₁, Zn₃ and Zn₂ respectively over that of Zn₀. A decrease in yields occurred by application of Zn and Fe at the high dose of 1000 mg L⁻¹ indicating a retarding effect at such high concentrations.

Bashir et al. (2016) stated that modulating Fe and Zn transport in chloroplast enhances plant growth development and metabolism. **Bastani et al. (2020)** reported positive response upon applying Fe in nano-form to tobacco plants (*Nicotiana tabacum*).

Table 2. Response of snap bean to Fe and Zn foliar spray: Pods fresh yield (kg ha⁻¹)

Zn fertilization (Zn)	Fe fertilization (Fe)				Mean
	Fe ₀	Fe ₁	Fe ₂	Fe ₃	
Zn ₀	998	933	934	841	821
Zn ₁	652	1018	1072	863	979
Zn ₂	993	802	1094	960	917
Zn ₃	978	848	1156	848	943
Mean	863	893	1028	878	

LSD 0.05 Zn: 31 ; Fe: 31 ; ZnFe: ns
 Notes: Fe₀, Fe₁, Fe₂ and Fe₃; also, Zn₀, Zn₁, Zn₂ and Zn₃ are solutions of (mg L⁻¹) 0, 250-nano, 500-nano and 1000 normal source respectively. Sprayed on days 37, 44, 51 and 58 after seeding

Dry yield of pods (Table 3)

The pattern of response to treatments resembled that concerning pod fresh yield. However plants receiving the high dose of both Zn and Fe (Zn₂Fe₂) gave the lowest yield of 67.9 kg ha⁻¹ while those receiving Zn₃Fe₂ gave the highest yield of 137.5 kg ha⁻¹ surpassing the lowest by 102.5%. The main effect of Fe shows lowest given by Fe₀ and highest given by Fe₂ with a pattern of Fe₂ > Fe₁ ≥ Fe₃ >

Fe₀ and increases averaging 19.0, 3.4 and 1.7% given by Fe₁, Fe₃ and Fe₂ respectively.

The main effect of Zn fertilization shows lowest given by Zn₀ and highest given by Zn₁ with a pattern of Zn₁ ≥ Zn₃ > Zn₂ > Zn₀; and increases averaging 19.2, 14.9 and 11.7% given by Zn₃, Zn₁ and Zn₂ respectively.

Table 3. Response of snap bean to Fe and Zn foliar spray: Pods dry yield (kg ha⁻¹)

Zn fertilization (Zn)	Fe fertilization (Fe)				Mean
	Fe ₀	Fe ₁	Fe ₂	Fe ₃	
Zn ₀	107.1	110.1	110.2	99.2	96.9
Zn ₁	67.9	120.1	126.5	101.8	115.5
Zn ₂	117.2	94.6	128.7	113.3	108.2
Zn ₃	115.4	100.1	137.5	100.1	111.3
Mean	101.9	105.4	121.3	103.6	

LSD 0.05 Zn: 4.5 ; Fe: 4.5 ; ZnFe: ns
 See footnotes of Table 2 for treatment designations

Shaban et al. (2019) noted positive response to nano Zn application to white beans (*Phaseolus vulgaris*). Therefore a balanced Zn/Fe combination is important for highest growth and productivity of bean pods (**Cakmak, 2008, Moghadam et al. 2012, Bashir et al. 2016 and Rudani and Prajapat 2018**). **Karimi (2014)** noted increased yields upon applying nano-Fe to mung beans (*Vigna radiata*). **Maralian (2010)** noted favourable response on growth and yield of wheat upon applying nano Fe and Zn. **Janmohammadi et al. (2016)** reported increased maize yield upon applying nano-nutrients as foliar spray. **Rose et al. (2019)** applied foliar zinc sulfate and zinc nanoparticles in coffee (*Coffea arabica* L.) and obtained a positive response.

Although Fe is an essential micronutrient, it may be toxic when in excess because it can react with oxygen which catalyzes formation of free radicals to oxidize organic molecules, thus leading to cell death (**Briat et al., 2010**)

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